

Common Polyatomic Ions—Learning-Laboratory.com

Practice on-screen or print. You may want to cross out ions not required by your instructor.

Name	Formula/Charge
Acetate *	$C_2H_3O_2^-$
Ammonium	NH_4^+
Carbonate	CO_3^{2-}
Chlorate	ClO_3^-
Chlorite	ClO_2^-
Chromate	CrO_4^{2-}
Cyanide	CN^-
Dichromate	$Cr_2O_7^{2-}$
Dihydrogen Phosphate	$H_2PO_4^-$
Hydrogen Carbonate (Bicarbonate)	HCO_3^-
Hydrogen Phosphate	HPO_4^{2-}
Hydrogen Sulfate (Bisulfate)	HSO_4^-
Hydrogen Sulfite (Bisulfite)	HSO_3^-
Hydronium	H_3O^+
Hydroxide	OH^-
Hypochlorite	ClO^-
Nitrate	NO_3^-
Nitrite	NO_2^-
Oxalate *	$C_2O_4^{2-}$
Perchlorate	ClO_4^-
Permanganate	MnO_4^-
Phosphate	PO_4^{3-}
Phosphite	PO_3^{3-}
Sulfate	SO_4^{2-}
Sulfite	SO_3^{2-}
Thiosulfate	$S_2O_3^{2-}$

* Organic